

#1

For panel A (Top) Two points on the line

$$\text{are } x (= {}^{27}\text{Al}/{}^{24}\text{Mg}) = 0 \quad y (= {}^{26}\text{Mg}/{}^{24}\text{Mg}) = 0.1393$$

$$x = 140 \quad y = 0.1438$$

$$\text{so } y \text{ intercept} = 0.1393$$

$$\text{slope} = \frac{\Delta y}{\Delta x} = \frac{0.1438 - 0.1393}{140} = 3.21 \times 10^{-5}$$

For panel B Two points on the line are

$$(x, y) = (0, 0.1390) \text{ and } (120, 0.145)$$

$$\text{so } y \text{ intercept} = 0.1390$$

$$\text{slope} = \frac{\Delta y}{\Delta x} = \frac{0.145 - 0.139}{120} = 5.0 \times 10^{-5}$$

The initial  $\frac{{}^{26}\text{Al}}{{}^{27}\text{Al}}$  ratios would be  $3.21 \times 10^{-5}$  and  $5.0 \times 10^{-5}$   
 The initial  $\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}}$  ratios would be 0.1393 and 0.1390

#2 In  $\delta$  notation with  $\left(\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}}\right)_{\text{std}} = 0.1394$ the initial  $\delta$  values would be

$$\text{A) } 1000 \times \left(\frac{0.1393}{0.1394} - 1\right) = -0.72$$

$$\text{B) } 1000 \times \left(\frac{0.1390}{0.1394} - 1\right) = -2.87$$

To show a more significant  $\delta$  value consider the  
 right most point in panel A, with  $\frac{{}^{26}\text{Mg}}{{}^{24}\text{Mg}} = 0.1427$

$$\text{For it, } \delta = 1000 \times \left(\frac{0.1427}{0.1394} - 1\right) = 23.67$$